

# Nickel-Catalyzed Cross-Coupling of Anisoles with Alkyl Grignard Reagents via C−O Bond Cleavage

Mamoru Tobisu,[\\*](#page-3-0),‡ Tsuyoshi Takahira,† and Naoto Chatani[\\*](#page-3-0),†

 $^\dagger$ Department of Applied Chemistry, Faculty of Engineering and  $^\ddagger$ Center for Atomic and Molecular Technologies, Graduate School of Engineering, Osaka University, Suita, Osaka 565-0871, Japan

**S** [Supporting Information](#page-2-0)

ABSTRACT: Nickel-catalyzed cross-coupling of methoxyarenes with alkyl Grignard reagents, which involves the cleavage of the C(aryl)−OMe bond, has been developed. The use of 1,3-dicyclohexylimidazol-2-ylidene as a ligand allows the introduction of a variety of alkyl groups, including Me,  $Me<sub>3</sub>SiCH<sub>2</sub>$ , ArCH<sub>2</sub>, adamantyl, and cyclopropyl. The method can also be used for the alkylative elaboration of complex molecules bearing a C(aryl)−OMe bond.



 $\sum$  ince the pioneerin[g](#page-3-0) works of Kumada and Tamao,<sup>[1](#page-3-0)</sup> and<br>Corriu and Masse,<sup>2</sup> the nickel-catalyzed cross-coupling of<br>and balides with Grimard reagents has continued to attract aryl halides with Grignard reagents has continued to attract considerable interest as a powerful method for the construction of carbon–carbon bonds.<sup>[3](#page-3-0)</sup> Despite the moisture/air sensitivity and modest functional group tolerance of Grignard reagents compared with the corresponding boron and zinc reagents, the Kumada−Tamao−Corriu (KTC) reaction remains a highly valued synthetic transformation. The popularity of this method can be attributed in part to the availability of a large number of inexpensive and nontoxic magnesium reagents, as evidenced by the many examples reported in the literature of the industrial application of this reaction. $<sup>4</sup>$  $<sup>4</sup>$  $<sup>4</sup>$  In addition to the practical</sup> advantages of this reaction, the high nucleophilicity of Grignard reagents can be used to allow the cross-coupling of unconventional coupling partners that would otherwise be too inert to react.<sup>[5](#page-3-0)−[9](#page-3-0)</sup> Among the unconventional groups reacted in this way, we were particularly intrigued by an early study reported by Wenkert describing the nickel-catalyzed KTC-type crosscoupling of methoxyarenes with  $ArMgX$ <sup>[10](#page-3-0)</sup> which involved the activation of an inert C(aryl)−OMe bond.[11](#page-3-0) Over the past decade, the scope of the nickel-catalyzed cross-coupling of methoxyarenes has expanded considerably to a wide range of nucleophiles, including organoboron,<sup>[12](#page-3-0)</sup> organozinc,<sup>[13](#page-3-0)</sup> organolithium,  $14$  hydride,  $15$  amine,  $16$  and boron<sup>[17](#page-3-0)</sup> nucleophiles. However, the scope of this reaction still remains limited compared with that of the corresponding cross-coupling reaction using aryl halides. In particular, the carbon nucleophiles that can be coupled with methoxyarenes have been primarily restricted to  $\overline{C}(\text{sp}^2)$ -based nucleophiles. We recently reported the first alkynylation of methoxyarenes, which allowed the introduction of a  $C(sp)$  center.<sup>[18](#page-3-0)</sup> With respect to their alkylation,<sup>[19](#page-3-0)</sup> methoxyarenes can only undergo crosscoupling with two specific  $C(sp^3)$  nucleophiles (Scheme 1). The first of these cross-coupling reactions is methylation using MeMgX,<sup>[20](#page-3-0)</sup> and the second involves the introduction of a trimethylsilylmethyl group using  $\rm Me_3SiCH_2Li.^{14a,21,22}$  $\rm Me_3SiCH_2Li.^{14a,21,22}$  $\rm Me_3SiCH_2Li.^{14a,21,22}$  $\rm Me_3SiCH_2Li.^{14a,21,22}$  $\rm Me_3SiCH_2Li.^{14a,21,22}$  Herein, we report that the 1,3-dicyclohexylimidazol-2-ylidene (ICy) ligand can be used to significantly expand the scope of the





alkylMgX reagents used in the nickel-catalyzed KTC-type crosscoupling of anisole derivatives. Notably, the use of the ICy ligand allowed the introduction of arylmethyl, adamantyl, and cyclopropyl groups via the cleavage of an Ar−OMe bond.

Since Dankwardt's seminal report on the KTC-type crosscoupling of anisoles with  $ArMgX$ <sup>[10c](#page-3-0)</sup> PCy<sub>3</sub> has been used as one of the most effective ligands for cross-coupling reactions involving the activation of the C(aryl)-O bonds in anisoles.<sup>[11](#page-3-0)</sup> We previously reported that NHC ligands could be used to achieve the nickel-catalyzed cross-coupling of methoxyarenes with nucleophiles that were otherwise unreactive when  $PCy_3$ was used as the ligand.<sup>[12c,15f,16,18](#page-3-0)</sup> These findings prompted us to investigate whether a series of NHC ligands could be used to promote the KTC-type cross-coupling of methoxyarenes with a wider range of Grignard reagents. We initially examined the effect of the ligand on the reaction of the anisole derivative 1 with  $Me<sub>3</sub>SiCH<sub>2</sub>MgCl$  (2a) using  $Ni(OAc)<sub>2</sub>$  as the catalyst precursor [\(Table 1](#page-1-0)). As expected, the use of  $PCy_3$  did not give any of the desired alkylated product 1a (entry 1). In contrast,

Received: July 29, 2015 Published: August 27, 2015 <span id="page-1-0"></span>Table 1. Ni-Catalyzed Cross-Coupling of 1 with 2a: the Effect of the Ligand<sup>a</sup>



<sup>a</sup>Reaction conditions: 1 (0.25 mmol), 2a (0.50 mmol),  $Ni(OAc)_{2}$ (0.0125 mmol), and ligand (0.025 mmol) in toluene (1.0 mL) at room temperature for 18 h.

the addition of an N-alkyl-substituted NHC to the reaction led to the formation of 1a with ICy being identified as the optimal NHC ligand in terms of the yield (entry 6). It should be noted that removing THF used during the preparation of  $Me<sub>3</sub>SiCH<sub>2</sub>MgCl$  and running the reaction in toluene alone is important for obtaining high conversion (see Supporting Information for details).

Having identified ICy as the optimal ligand, we proceeded to examine the scope of the Grignard reagents in the nickelcatalyzed cross-coupling of methoxyarene 3 (Table 2). Given that the alkyl Grignard reagents bearing a  $\beta$ -hydrogen, such as  $C_5H_{11}MgBr$  (2j, entry 10) and 'PrMgBr (2k, entry 11), afforded the undesired reduction product (i.e., naphthalene)<sup>[23](#page-3-0)</sup> instead of the alkylation product, we focused specifically on Grignard reagents with no  $\beta$ -hydrogens.  ${}^t$ BuCH<sub>2</sub>MgBr (2b) behaved in a similar manner to 2a and efficiently underwent the cross-coupling at room temperature to provide 3b (entry 2). This result demonstrated that the presence of an  $\alpha$ -silyl group in the Grignard reagent, as in 2a, was not critical to the success of this alkylation process. The methylation (entry 3) and benzylation (entry 4) reactions also proceeded smoothly using the Ni/ICy system, albeit at a higher reaction temperature of 100 °C. Sterically hindered (entry 5) and  $\pi$ -extended ArCH<sub>2</sub> (entry 6) groups were also amenable to this reaction, allowing the synthesis of various diarylmethane derivatives from readily available methoxyarenes. Two adamantyl-Grignard reagents, 2g and 2h, were found to be suitable nucleophiles for the nickelcatalyzed cross-coupling of 3, even though they both have  $\beta$ hydrogen atoms. These arylations successfully occurred at both the secondary (entry 7) and tertiary (entry 8) carbon atoms of the adamantyl group, most likely because the undesired  $\beta$ hydrogen elimination would be forbidden by Bredt's rule.<sup>24</sup> These results therefore represent the first reported examples of the cross-coupling reactions of methoxyarenes with secondary and tertiary alkyl groups. It is noteworthy that cyclopropyl-MgBr (2i, entry 9) was also coupled with 3 under these nickelcatalyzed conditions to form 2-cyclopropylnaphthalene (3i), along with a ring opened product  $3i'$  ( $3i:3i' = 9:1$ ).

As shown in [Scheme 2,](#page-2-0) the Ni/ICy catalyst catalyzed the KTC-type cross-coupling of a wide variety of methoxyarenes. Notably, these reaction conditions could be applied to ethoxy and isopropoxy groups as well as methoxy groups. Consistent with the previous cross-couplings of methoxyarenes, $11$  a C(naphthyl)−O bond is more reactive than a C(phenyl)−O bond, allowing the regioselective alkylation of 2-phenoxynaph-



Table 2. Ni/ICy-Catalyzed Alkylation of Aryl Methyl Ethers:

Scope of the Grignard Reagents<sup> $a$ </sup>

a Reaction conditions: 3 (0.25 mmol), Grignard reagent (0.50 mmol),  $Ni(OAc)_2$  (0.0125 mmol), and ICy·HCl (0.025 mmol) in toluene (1.0 mL) for 18 h. <sup>b</sup>Isolated yield of the alkylated product unless otherwise noted. "The yield was determined by GC due to the volatility of the product.  ${}^{d}$ The reaction was run using  $Ni(OAc)_2$  (0.025 mmol) and ICy·HCl (0.050 mmol). <sup>e</sup> The yield was determined by NMR analysis. The product was formed as a 9:1 mixture of 2-cyclopropylnaphthalene (3i) and  $(E)$ -2-(prop-1-en-1-yl)naphthalene  $(3i')$ .  $^{f}$ Ni(OAc)<sub>2</sub> (0.050) mmol) and ICy·HCl  $(0.050 \text{ mmol})$  were used. <sup>g</sup>GC yield of naphthalene.

thalene to form 3a. Functional groups, including amines 8, acetals 12, and amides 13, were well tolerated. One useful feature of this KTC-type cross-coupling of methoxyarenes is its compatibility with a free hydroxyl group in the substrate by simply increasing the number of equivalents of the Grignard reagents to 3 equiv, as exemplified by the reactions of 9, 10, and 11. This protocol was also found to be robust enough for the alkylation of the methoxy group at sterically congested positions, such as those found in compounds 7 and 14. Moreover, several heteroaryl ethers, including pyridine 16, quinoline 17, carbazole 18, and indole 19, were successfully alkylated under these conditions via the loss of their methoxy groups. This alkylation method was also applicable to alkenyl ether substrates. For example, a cis-substituted homoallylic alcohol was synthesized by the cross-coupling of 2,3 dihydrofuran 21. It is noteworthy that this alkylation can be performed on the gram scale using 1 mol % of the catalyst [\(eq](#page-2-0)  $1).^{25}$  $1).^{25}$  $1).^{25}$  $1).^{25}$ 

We also examined alkylation of 1,2-dimethoxybenzene (22) [\(Table 3\)](#page-2-0). When 2a (entries 1 and 2) or 2b (entries 3 and 4) was used as the alkylating reagent in the Ni/ICy-catalyzed reaction of 22, the monoalkylated product was produced selectively with the second methoxy group in the product remaining intact, even when a large excess (4.0 equiv) of the

<span id="page-2-0"></span>

a Reaction conditions: aryl ether (0.25 mmol), Grignard reagent (0.50 mmol),  $Ni(OAc)_{2}$  (0.0125 mmol), and ICy·HCl (0.025 mmol) in toluene (1.0 mL) for 18 h. The yields shown are the isolated yield. b Grignard reagent (1.0 mmol) was used. 'Grignard reagent (0.75 mmol) was used.  $Ni(OAc)_2$  (0.025 mmol) and ICy·HCl (0.050 mmol) were used.  $e^{\tan \theta}$  at 80  $^{\circ}$ C. The yield was determined by GC because of the volatility of the product.  ${}^{8}Ni(OAc)_{2}$  (0.0075 mmol) and ICy·HCl (0.015 mmol) were used.



Grignard reagent was used. In contrast, the use of 2d as an alkylating agent allowed a high level of control over the mono/ di alkylation depending on the amount of the Grignard reagent used (entries 5 and 6). These results can be rationalized by a simple steric effect with the  $\text{Me}_3\text{SiCH}_2$  and  $\text{fBuCH}_2$  (Charton  $\nu$ value:  $1.34)^{26}$  $1.34)^{26}$  $1.34)^{26}$  groups being large enough to block a second alkylation at their ortho position. In contrast, the  $PhCH<sub>2</sub>$  group (Charton  $\nu$  value: 0.70) would be too small to block the second alkylation.

The power and utility of this alkylation method was further demonstrated by its application to complex methoxyarene substrates (Scheme 3). For example, the structural elaboration of complex alkaloid dextromethorphan 23, which is a marketed antitussive agent, was readily accomplished using our nickel-



a The yield was determined by GC because of the volatility of the product.





catalyzed KTC-type cross-coupling. Furthermore, the alkylation of estradiol proceeded smoothly using the corresponding methyl ether 24 while leaving the aliphatic alcohol moiety intact.

In summary, we have shown that a Ni/ICy system can be used to promote the KTC-type alkylative cross-coupling of methoxyarenes with a range of alkyl Grignard reagents. This new protocol allows, for the first time, the introduction of ArCH<sub>2</sub>, adamantyl, and cyclopropyl groups to methoxyarenes via the cleavage of their C(aryl)−O bonds. Considering the widespread availability and robust nature of compounds bearing a methoxy group, this new alkylation method could be used for the late stage elaboration of complex aryl ethers.

## ■ ASSOCIATED CONTENT

### **S** Supporting Information

The Supporting Information is available free of charge on the ACS Publications website at DOI: 10.1021/acs.orglett.5b02200.

<span id="page-3-0"></span>Detailed experimental procedures and characterization of products (PDF)

# ■ AUTHOR INFORMATION

#### Corresponding Authors

\*E-mail: tobisu@chem.eng.osaka-u.ac.jp \*E-mail: chatani@chem.eng.osaka-u.ac.jp

# Notes

The authors declare no competing financial interest.

# ■ ACKNOWLEDGMENTS

This work was supported by a Grant-in-Aid for Scientific Research on Innovative Areas "Molecular Activation Directed toward Straightforward Synthesis" from MEXT, Japan, and ACT-C from JST, Japan. We also thank the Instrumental Analysis Center, Faculty of Engineering, Osaka University, for their assistance with HRMS.

### **ENDERGERIERS**

(1) Tamao, K.; Sumitani, K.; Kumada, M. J. Am. Chem. Soc. 1972, 94, 4374.

(2) Corriu, R. J. P.; Masse, J. P. J. Chem. Soc., Chem. Commun. 1972, 144a.

(3) A review: (a) Banno, T.; Hayakawa, Y.; Umeno, M. J. Organomet. Chem. 2002, 653, 288. Selected recent examples: (b) Zacuto, M. J.; Shultz, C. S.; Journet, M. Org. Process Res. Dev. 2011, 15, 158. (c) Gangula, S.; Neelam, U. K.; Baddam, S. R.; Dahanukar, V. H.; Bandichhor, R. Org. Process Res. Dev. 2015, 19, 470.

(4) Selected reviews of KTC cross-coupling: (a) Tamao, K. J. Organomet. Chem. 2002, 653, 23. (b) Adrio, J.; Carretero, J. C. ChemCatChem 2010, 2, 1384.

(5) Selected examples of the KTC-type reaction of aryl fluorides: (a) Kiso, Y.; Tamao, K.; Kumada, M. J. Organomet. Chem. 1973, 50, C12. (b) Bö hm, V. P. W.; Gstöttmayr, C. W. K.; Weskamp, T.; Herrmann, W. A. Angew. Chem., Int. Ed. 2001, 40, 3387. (c) Mongin, F.; Mojovic, L.; Guillamet, B.; Trécourt, F.; Quéguiner, G. J. Org. Chem. 2002, 67, 8991. (d) Terao, J.; Ikumi, A.; Kuniyasu, H.; Kambe, N. J. Am. Chem. Soc. 2003, 125, 5646. (e) Dankwardt, J. W. J. Organomet. Chem. 2005, 690, 932. (f) Ackermann, L.; Born, R.; Spatz, J. H.; Meyer, D. Angew. Chem., Int. Ed. 2005, 44, 7216. (g) Yoshikai, N.; Mashima, H.; Nakamura, E. J. Am. Chem. Soc. 2005, 127, 17978. (h) Yoshikai, N.; Matsuda, H.; Nakamura, E. J. Am. Chem. Soc. 2009, 131, 9590. (i) Wang, J.-R.; Manabe, K. Org. Lett. 2009, 11, 741.

(6) A review on the KTC-type reaction of phenol derivatives: (a) Li, W.-N.; Wang, Z.-L. RSC Adv. 2013, 3, 25565. More general reviews on C(aryl)−O activation: (b) Yu, D.-G.; Li, B.-J.; Shi, Z.-J. Acc. Chem. Res. 2010, 43, 1486. (c) Li, B.-J.; Yu, D.-G.; Sun, C.-L.; Shi, Z.-J. Chem. - Eur. J. 2011, 17, 1728. (d) Rosen, B. M.; Quasdorf, K. W.; Wilson, D. A.; Zhang, N.; Resmerita, A.-M.; Garg, N. K.; Percec, V. Chem. Rev. 2011, 111, 1346. (e) Mesganaw, T.; Garg, N. K. Org. Process Res. Dev. 2013, 17, 29. (f) Tobisu, M.; Chatani, N. Top. Organomet. Chem. 2013, 44, 35. (g) Yamaguchi, J.; Muto, K.; Itami, K. Eur. J. Org. Chem. 2013, 2013, 19. (h) Han, F.-S. Chem. Soc. Rev. 2013, 42, 5270.

(7) General reviews on C−S activation: (a) Wang, L.; He, W.; Yu, Z. Chem. Soc. Rev. 2013, 42, 599. (b) Pan, F.; Shi, Z.-J. ACS Catal. 2014, 4, 280.

(8) The KTC-type reaction of aryl cyanides: (a) Miller, J. A. Tetrahedron Lett. 2001, 42, 6991. (b) Miller, J. A.; Dankwardt, J. W. Tetrahedron Lett. 2003, 44, 1907.

(9) Selected examples of cross-coupling of C−H bonds with Grignard reagents: (a) Ilies, L.; Asako, S.; Nakamura, E. J. Am. Chem. Soc. 2011, 133, 7672. (b) Chen, Q.; Ilies, L.; Yoshikai, N.; Nakamura, E. Org. Lett. 2011, 13, 3232. (c) Yoshikai, N.; Asako, S.; Yamakawa, T.; Ilies, L.; Nakamura, E. Chem. - Asian J. 2011, 6, 3059. (10) The KTC-type cross-coupling of methoxyarenes with ArMgX:

(a) Wenkert, E.; Michelotti, E. L.; Swindell, C. S. J. Am. Chem. Soc.

1979, 101, 2246. (b) Wenkert, E.; Michelotti, E. L.; Swindell, C. S.; Tingoli, M. J. Org. Chem. 1984, 49, 4894. (c) Dankwardt, J. W. Angew. Chem., Int. Ed. 2004, 43, 2428. (d) Xie, L.-G.; Wang, Z.-X. Chem. - Eur. J. 2011, 17, 4972. (e) Zhao, F.; Yu, D.-G.; Zhu, R.-Y.; Xi, Z.; Shi, Z. J. Chem. Lett. 2011, 40, 1001. (f) Iglesias, M. J.; Prieto, A.; Nicasio, M. C. Org. Lett. 2012, 14, 4318. (g) Zhao, F.; Zhang, Y.-F.; Wen, J.; Yu, D.- G.; Wei, J.-B.; Xi, Z.; Shi, Z.-J. Org. Lett. 2013, 15, 3230. (h) Cornella, J.; Martin, R. Org. Lett. 2013, 15, 6298.

(11) Reviews on catalytic reactions involving the cleavage of C(aryl)−OMe bonds: (a) Cornella, J.; Zarate, C.; Martin, R. Chem. Soc. Rev. 2014, 43, 8081. (b) Tobisu, M.; Chatani, N. Acc. Chem. Res. 2015, 48, 1717.

(12) (a) Tobisu, M.; Shimasaki, T.; Chatani, N. Angew. Chem., Int. Ed. 2008, 47, 4866. (b) Shimasaki, T.; Konno, Y.; Tobisu, M.; Chatani, N. Org. Lett. 2009, 11, 4890. (c) Tobisu, M.; Yasutome, A.; Kinuta, H.; Nakamura, K.; Chatani, N. Org. Lett. 2014, 16, 5572.

(13) Wang, C.; Ozaki, T.; Takita, R.; Uchiyama, M. Chem. - Eur. J. 2012, 18, 3482.

(14) (a) Leiendecker, M.; Hsiao, C.-C.; Guo, L.; Alandini, N.; Rueping, M. Angew. Chem., Int. Ed. 2014, 53, 12912. (b) Guo, L.; Leiendecker, M.; Hsiao, C.-C.; Baumann, C.; Rueping, M. Chem. Commun. 2015, 51, 1937.

(15) (a) Á lvarez-Bercedo, P.; Martin, R. J. Am. Chem. Soc. 2010, 132, 17352. (b) Tobisu, M.; Yamakawa, K.; Shimasaki, T.; Chatani, N. Chem. Commun. 2011, 47, 2946. (c) Sergeev, A. G.; Hartwig, J. F. Science 2011, 332, 439. (d) Cornella, J.; Gómez-Bengoa, E.; Martin, R. J. Am. Chem. Soc. 2013, 135, 1997. (e) Sergeev, A. G.; Webb, J. D.; Hartwig, J. F. J. Am. Chem. Soc. 2012, 134, 20226. See also: (f) Tobisu, M.; Morioka, T.; Ohtsuki, A.; Chatani, N. Chem. Sci. 2015, 6, 3410.

(16) (a) Tobisu, M.; Shimasaki, T.; Chatani, N. Chem. Lett. 2009, 38, 710. (b) Tobisu, M.; Yasutome, A.; Yamakawa, K.; Shimasaki, T.; Chatani, N. Tetrahedron 2012, 68, 5157.

(17) Zarate, C.; Manzano, R.; Martin, R. J. Am. Chem. Soc. 2015, 137, 6754.

(18) Tobisu, M.; Takahira, T.; Ohtsuki, A.; Chatani, N. Org. Lett. 2015, 17, 680.

(19) A review on cross-coupling using alkyl-organometallic reagents: Jana, R.; Pathak, T. P.; Sigman, M. S. Chem. Rev. 2011, 111, 1417.

(20) Guan, B.-T.; Xiang, S.-K.; Wu, T.; Sun, Z.-P.; Wang, B.-Q.; Zhao, K.-Q.; Shi, Z.-J. Chem. Commun. 2008, 1437.

(21) Benzyl and 2-phenethyl groups can be introduced to methoxyarenes bearing an ortho directing group under ruthenium catalysis: Kakiuchi, F.; Usui, M.; Ueno, S.; Chatani, N.; Murai, S. J. Am. Chem. Soc. 2004, 126, 2706.

(22) The non-catalytic alkylation of methoxyarenes bearing an ortho carbonyl group with RMgX has been reported to occur via an  $S_N$ Ar mechanism: Jiménez-Osés, G.; Brockway, A. J.; Shaw, J. T.; Houk, K. N. J. Am. Chem. Soc. 2013, 135, 6633 and references therein..

(23) The reduction product is most likely to be generated through undesired β-hydrogen elimination of the alkylnickel intermediate rather than the nickel methoxide intermediate (see ref 15f) based on a labeling experiment. See the Supporting Information for details.

(24) Shea, K. J. Tetrahedron 1980, 36, 1683.

(25) Additional notes regarding the scope and limitations: (a) Benzyl methyl ether was unreactive under these conditions. (b) Acetal and amide functionalities reacted when the reaction was performed at 100  $^{\circ}C.$ 

(26) Charton, M. J. Am. Chem. Soc. 1975, 97, 1552.